# organic compounds

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# 4'-Vinyl-2,2':6',2"-terpyridine

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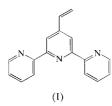
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The title compound,  $C_{17}H_{13}N_3$ , is a versatile precursor for polymeric terpyridine derivatives and their metal complexes. The molecule has transoid and near-coplanar pyridine rings. However, the vinyl group is forced out of the plane of the terpyridyl moiety by a close  $H \cdots H$  contact.

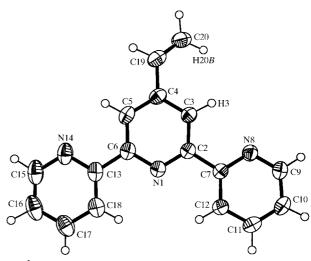
#### Comment

4'-Vinyl-2,2':6',2"-terpyridine (Potts & Konwar, 1991) is a versatile synthetic intermediate, which undergoes homo- or copolymerization reactions under free-radical or electrochemical conditions. The metal-containing polymers obtained in this way have potentially interesting photophysical or electrocatalytic properties (Potts et al., 1987; Guadalupe et al., 1988; Potts & Usifer, 1988; Hurrell et al., 1989). We have obtained this compound, (I), as a synthetic intermediate during our studies of complexes of substituted meridional trisimine ligands (Solanki et al., 1998, 1999), and now report its crystal structure (Fig. 1).



As with other 2,2':6',2"-terpyridine derivatives (Constable et al., 1990, 1995; Constable, Cargill Thompson et al., 1992; Constable, Kahn et al., 1992; Bessel et al., 1992; Fallahpour et al., 1999), the three pyridine rings are transoid to each other and are nearly coplanar, the dihedral angles between the leastsquares planes of the three rings being N1-C6/C7-C12 = $7.1 (2)^{\circ}$  and N1–C6/C13–C18 = 2.2 (2)°. The torsion angles about the C4-C19 bond are C5-C4-C19-C20 =  $-155.28 (17)^{\circ}$  and C3-C4-C19-C20 = 24.4 (3)°, so that the vinyl substituent is not coplanar with the central pyridine ring. This can be attributed to the effect of steric repulsions between the H atoms H3 and H20B, which lie only 2.35 Å apart.

In the lattice, the molecules are arranged into  $\pi$ - $\pi$ -stacked pairs related by inversion (symmetry code: -x, 1 - y, -z).





Molecular structure showing 50% probability displacement ellipsoids and the atom-numbering scheme employed. H atoms have arbitrary radii and those referred to in the text are labelled.

This leads to a  $\pi$ - $\pi$  interaction between pyridine rings N1–C6 and C7-C12 of one molecule, and N1'-C6' and C7'-C12' of a neighbouring molecule. The average interplanar spacing for this interaction is 3.42 Å, the dihedral angle between the leastsquares planes of the two rings is  $7.1 (2)^{\circ}$  and the offset of their centroids is 2.15 Å (Hunter & Sanders, 1990). These pairs of molecules are in turn arranged translationally into onedimensional chains parallel to the *a* axis, leading to a second  $\pi$ - $\pi$  interaction between C7-C12 and C7''-C12'' of molecules related by another inversion centre (1 - x, 1 - y, -z). These rings are therefore strictly coplanar by symmetry, their interplanar spacing being 3.42 Å while their centroids are offset by 1.43 (2) Å. Adjacent  $\pi$ - $\pi$  chains are related by the operation of the *c*-glide  $\left(-\frac{1}{2} - x, -\frac{1}{2} + y, z\right)$  and are rotated by 61.9 (2)° with respect to each other.

## **Experimental**

The compound was isolated in moderate yield from the reaction of 4'-(2-hydroxyethyl)-2,2':6',2"-terpyridine (Potts *et al.*, 1987) with SeO<sub>2</sub> in refluxing dioxane. The NMR spectra of the resultant white solid matched those previously reported for this compound (Potts et al., 1987). Crystals were grown from ether: hexane (1/1).

Crystal data	
$C_{17}H_{13}N_3$ $M_r = 259.30$ Orthorhombic, <i>Pbca</i> a = 7.9476 (2) Å b = 11.5546 (3) Å c = 29.4257 (8) Å V = 2702.20 (12) Å <sup>3</sup> Z = 8 $D_x = 1.275$ Mg m <sup>-3</sup>	Mo K $\alpha$ radiation Cell parameters from 14681 reflections $\theta = 2.77-27.49^{\circ}$ $\mu = 0.078 \text{ mm}^{-1}$ T = 150 (2)  K Plate, colourless $0.67 \times 0.50 \times 0.25 \text{ mm}$
Data collection Nonius KappaCCD area-detector diffractometer Area-detector scans Absorption correction: multi-scan (SORTAV; Blessing, 1995) $T_{\min} = 0.950, T_{\max} = 0.981$ 14681 measured reflections	3093 independent reflections 2246 reflections with $I > 2\sigma(I)$ $R_{int} = 0.074$ $\theta_{max} = 27.49^{\circ}$ $h = -10 \rightarrow 10$ $k = -14 \rightarrow 15$ $l = -38 \rightarrow 38$

Refinement

 $\begin{array}{ll} \mbox{Refinement on } F^2 & w = 1/[\sigma^2(F_o^2) + (0.0781P)^2 \\ R[F^2 > 2\sigma(F^2)] = 0.052 & + 0.3186P] \\ wR(F^2) = 0.147 & where P = (F_o^2 + 2F_c^2)/3 \\ S = 1.044 & (\Delta/\sigma)_{max} < 0.001 \\ 3093 \ reflections & \Delta\rho_{max} = 0.41 \ e \ {\rm \AA}^{-3} \\ 233 \ parameters & \Delta\rho_{min} = -0.20 \ e \ {\rm \AA}^{-3} \end{array}$ 

All H atoms were located in a Fourier difference map, and were allowed to refine freely. The C-H bond lengths are in the range 0.951 (16)–1.03 (2) Å.

Data collection: *COLLECT* (Nonius, 1999); cell refinement: *DENZO-SMN* (Otwinowski & Minor, 1996); data reduction: *DENZO-SMN*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEX* (McArdle, 1995); software used to prepare material for publication: local program.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: BM1415). Services for accessing these data are described at the back of the journal.

## References

- Bessel, C. A., See, R. F., Jameson, D. L., Churchill, M. R. & Takeuchi, K. J. (1992). J. Chem. Soc. Dalton Trans. pp. 3223–3228.
- Blessing, R. H. (1995). Acta Cryst. A51, 33-38.
- Constable, E. C., Cargill Thompson, A. M. W., Harveson, P., Macko, L. & Zehnder, M. (1995). *Chem. Eur. J.* 1, 360–367.
- Constable, E. C., Cargill Thompson, A. M. W., Tocher, D. & Daniels, M. A. M. (1992). *New J. Chem.* **16**, 855–867.
- Constable, E. C., Kahn, F. K., Raithby, P. R. & Marquez, V. E. (1992). Acta Cryst. C48, 932–934.
- Constable, E. C., Lewis, J., Liptrot, M. C. & Raithby, P. R. (1990). *Inorg. Chim. Acta*, **178**, 47–54.
- Fallahpour, R.-A., Neuburger, M. & Zehnder, M. (1999). New J. Chem. 23, 53–61.
- Guadalupe, A. R., Usifer, D. A., Potts, K. T., Hurrell, H. C., Mogstad, A.-E. & Abruna, H. D. (1988). J. Am. Chem. Soc. 110, 3462–3466.
- Hunter, C. A. & Sanders, J. K. M. (1990). J. Am. Chem. Soc. 112, 5525–5534. Hurrell, H. C., Mogstad, A.-L., Usifer, D. A., Potts, K. T. & Abruna, H. D.
- (1989). Inorg. Chem. 28, 1080–1084.
- McArdle, P. (1995). J. Appl. Cryst. 28, 65.
- Nonius (1999). COLLECT. Nonius BV, Delft, The Netherlands.
- Otwinowski, Z. & Minor, W. (1996). Methods Enzymol. 276, 307-326.
- Potts, K. T. & Konwar, D. (1991). J. Org. Chem. 56, 4815-4816.
- Potts, K. T. & Usifer, D. A. (1988). Macromolecules, 21, 1985–1991.
- Potts, K. T., Usifer, D. A., Guadalupe, A. & Abrams, H. D. (1987). J. Am. Chem. Soc. 109, 3961–3967.
- Sheldrick, G. M. (1990). Acta Cryst. A46, 467-473.
- Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.
- Solanki, N. K., McInnes, E. J. L., Mabbs, F. E., Radojevic, S., McPartlin, M., Feeder, N., Davies, J. E. & Halcrow, M. A. (1998). Angew. Chem. Int. Ed. Engl. 37, 2221–2223.
- Solanki, N. K., Wheatley, A. E. H., Radojevic, S., McPartlin, M. & Halcrow, M. A. (1999). J. Chem. Soc. Dalton Trans. pp. 521–523.